

## **THE OCCURRENCE AND CRYSTAL STRUCTURE OF TUNGSTEN BEARING ORES-MINERAL AT KANBOUK, TANINTHARYI REGION (MYANMAR)**

Min Maung Maung<sup>1</sup>, Sint Ohnmar<sup>2</sup>

### **Abstract**

Tungsten is a metal of superlatives and the lowest vapour pressure and the lowest expansion coefficient of all metals. The quality and the costs of producing the metal greatly depends on the efficiency of the concentrating steps performed at the rather limited number of mining sites. In this work, the tungsten (W) and tin (Sn) were produced in wolframite ore from Kanboug mine by calcinations process and their structural properties were analyzed by using X-ray diffraction method. The elemental analysis was done by Energy Dispersive X-rays Fluorescence (EDXRF) method. The electric and magnetic properties of these samples were also observed PERMAGRAPH L and LCR meter.

**Keywords:** Tungsten, LCR meter, Wolframite, X-ray diffraction

### **Introduction**

Tungsten ore is a rock from which the element tungsten can be economically extracted. The ore minerals of tungsten include wolframite, scheelite, and ferberite. Materials processing is one of the most important and active areas of research in heat transfer today. With growing international competition, it is has become crucial to improve the present processing techniques and the quality of the final product. Heat transfer is extremely important in a wide range of materials processing techniques such as crystal growing, casting, glass fiber drawing, chemical vapor deposition, spray coating, soldering, welding, polymer extrusion, injection molding, and composite materials fabrication.

The flows that arise in the molten material in crystal growing due to temperature and concentration differences, for instance, can affect the quality of the crystal and, thus, of the semiconductors fabricated from the crystal. Therefore, it is important to understand these flows and develop methods to minimize or control their effects. As a consequence of the importance of heat

<sup>1</sup> Dr, Associate Professor, Department of Physics, Dawei University.

<sup>2</sup> Dr, Lecturer, Department of Physics, Dawei University.

and mass transfer in materials processing, extensive work is presently being directed at this area. But what is often lacking is the link between the basic mechanisms that govern diverse processing techniques and the thermal systems needed to achieve the given process.

Tungsten is an economically important metal, being widely used in light-bulb filaments, electron and television tubes, abrasives and special alloys such as tool steels. Tungsten carbide is of great importance to the metal-working, mining and petroleum industries. Contamination from these sources is, therefore, possible in industrial and urban areas.

### **Experimental**

Rock samples were purchased from Wolframite mine (Kanbouk), Tanitharyi region. Firstly, tungsten-ores alloys were divided and collected from rock sample according to their raw group formations (colour). The obtained raw samples were grounded with agate mortar and heated at 150°C, 300°C, 400°C and 500°C in furnace 3 hrs each and slightly cold down to room temperature. The samples were grounded again and again to form homogeneous powder. The powder samples were then added into the internal mixer 3 min after the blending started and allowed to mix for 4 min and were charged into the internal mixer. The powder samples were prepared with XRD holder to determine the structure and lattice parameter.

Small amount of annealed powder was characterized by XRD using Cu-K $\alpha$  radiation at room temperature in order to check the phase purity and calculated average crystallite size. Once a homogeneous mixture is assumed after 13 min, the samples were pressed by hydraulic press machine into a measurement cell (1.3 cm diameter). The pellet samples were cured in a mold with a spacer approximately 0.5 mm thick and 1.3 cm in diameter. Dielectric measurements were carried out by a digital RLC meter (GWInstek LCR821) at room temperature. The dielectric constant and loss factor was measured over the frequency range 10 to 200 kHz. The measurements were performed applying a voltage of 1.5 V rms between the electrodes no. 1 and electrode no. 2. Capacitive and resistance effects were determined through the measurements of permittivity and resistivity. The room temperature

ferroelectric hysteresis loop measurement demonstrated the coexistence of ferroelectricity and magnetism was done by permagraph L equipment.

## **Results and Discussion**

### **Structural characterization and elemental concentration**

Figure 1.1 showed the XRD (X-ray diffraction) patterns of the powder samples. It was found that the powders were in tetragonal structure having  $p4_2/m\ n\ m$  space group. The average lattice constants are calculated to be  $a = 4.737\text{\AA}$  and  $c = 3.185\ \text{\AA}$  from the refinement of the XRD data. The prominent peaks in the plot are indexed to various (hkl) planes of Cassiterite (Tin(IV)oxide ( $\text{SnO}_2$ )). The secondary peaks were observed in the form of wolframite ( $\text{Fe, MnWO}_4$ ). The sample calcined at  $400^\circ\text{C}$  and  $500^\circ\text{C}$  are having stronger peak of (211) plane compared with other samples, but there have more other planes and impurities.

From the elemental analysis of wolframite ore samples, Tin (Sn), Tungsten (W), Iron (Fe), Manganese (Mn) were observed as major elements. When increasing of the treatment of annealing temperature, the concentration of Tin (Sn) element was decreased and the concentration of Tungsten (W) and Iron (Fe) were increased. Table 1.1 was shown the elemental concentrations of wolframite ore samples with heat treatment.

### **Surface morphology analysis**

Figure 1.2 showed the scanning electron microscope (SEM) images of wolframite ore samples with different annealing temperatures. It was analyzed in a JEOL JSM-6400 scanning electron microscope at accelerating voltage of 20KVA, real time of 21-36 and live time of 60 seconds. There were two different types of morphological feathers along with voids are visible. The unreacted  $\text{MnWO}_4$  grains which exhibit flowery feathers depict breaking up morphology, evenly distributed on the whole pellet surface. The increase in the furnace temperature enhances the rate of reaction and the images were formed faceted particles and agglomerated rods.

### **Dielectric measurements at room temperature**

Figure 1.3 showed the dielectric constant measurements with respect to frequency in the region of (10 Hz-200 kHz) of wolframe pellets produced using PVA as a binder and applying pressure of 15 ton/cm<sup>2</sup> and using LCR meter. It was found that dielectric constant was higher in the lower frequency region. It decreases with increasing frequency and becomes almost constant at higher frequency region. It can be seen that the dielectric constant decreases with increasing frequency. It means that the dielectric constant of these samples were the strong function of frequency. The average dielectric constant of wolframe samples in the range 10Hz to 200 kHz at room temperature found to be  $\epsilon_r = 35, 30, 25, 20, 10$  respectively.

### **Resistivity measurement at room temperature**

Figure 1.4 showed the electrical resistivity decreases and their electrical conductivity increases with increasing frequency applied. It can be said that the possibility of uses to these materials in high frequency device applications.

### **Magnetic hysteresis loop measurement**

Figure 1.5 showed the magnetization of the wolframe simples as a function of applied magnetic field. Magnetization and magnetic hysteresis results confirm the absence of canted ferromagnetic behaviour in these simples. The measurement was carried out on the simples at room temperature. It is evident that magnetization is a linear function of applied magnetic field. Their magnetic remanences, normal coercivity and maximum energy products were decreased but their intrinsic coercivity values were increased with increasing annealing temperatures. It was said that this behaviour is the typical of antiferromagnetic materials.

### **Conclusion**

Although the availability and quality of tungsten ores are limited and the price of the metal greatly fluctuates, its special physical properties and applications pose a reliable demand for production. The low tungsten content

and the interference of the gangue minerals in the ore make it complicated and costly to provide a suitable concentrate for the metallurgical processing.

The primary objective of this work was the characterization of an ore bearing wolfrinite mineral from Kanbouk mine, Tanintharyi region. The XRD phase patterns confirmed the availability of minerals such as Cassiterite, Hubnerite and manganocolumbite. The chemical elemental composition determined by EDXRF was Tin (Sn), Tungsten (W), Iron (Fe), Manganese (Mn) as major elements and other trace elements were also found. The mineralogical studies carried out with SEM point imaging showed the presence of different aggregates of minerals.

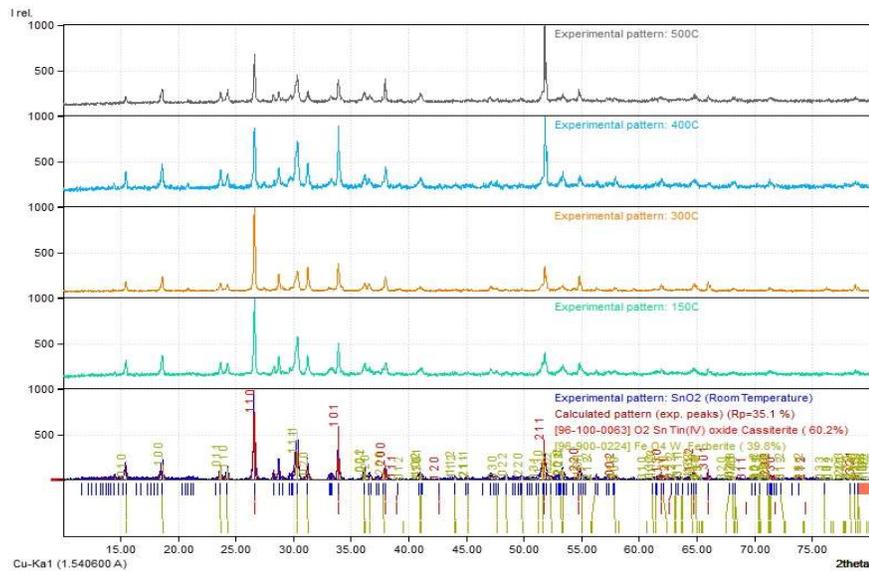
A decrease in dielectric constant with increase of frequency was observed. This decrease was rapid at lower frequency ranges and slower at higher frequency ranges. It is observed that the dielectric structure is formed into two layers namely, the well conducting grains and poorly conducting grain boundaries. At very low frequencies, the oxide grain boundaries were more active and this contributes to the very high dielectric constant at low frequencies. Thus, it was found that they were a promising material for dielectric applications. B-H loop confirms that the antiferromagnetic nature of the wolframite samples. Since the calculation values of permeability of these samples, it was said that their magnetic properties would transformed from diamagnetism to paramagnetism in sample annealing with temperature above 500°C.

**Table 1.1:** The elemental concentration in wolframite ore samples.

Elements	Concentration (%)				
	room temperature	Calcinations for 1 hour			
		150°C	300°C	400°C	500°C
Tin (Sn)	34.251	44.331	31.878	26.737	25.965
Tungsten (W)	26.949	25.390	26.330	26.897	28.353
Iron (Fe)	30.21	22.644	32.290	36.015	36.082
Manganese (Mn)	7.053	5.402	6.989	7.298	7.530

**Table 1.2:** The characteristic quantities of magnetic field in wolframite ore samples.

Sample	Remanence ( $B_r$ )(T)	Intrinsic coercivity ( $H_{cJ}$ ) (kA/m)	Normal coercivity ( $H_{cB}$ ) (kA/m)	Maximum Energy Product $(BH)_{max}$ ( $\text{kJ/m}^3$ )
Room Temperature	0.00391	11.6	3.2	0.00032
Annealing with 150°C	0.00349	37.6	2.89	0.00009
Annealing with 300°C	0.00237	66.4	1.91	0.0011
Annealing with 400°C	0.00196	72.08	1.57	0.00114
Annealing with 500°C	0.000951	90.9	0.749	0.00001

**Figure 1.1:** The XRD patterns of wolframite samples

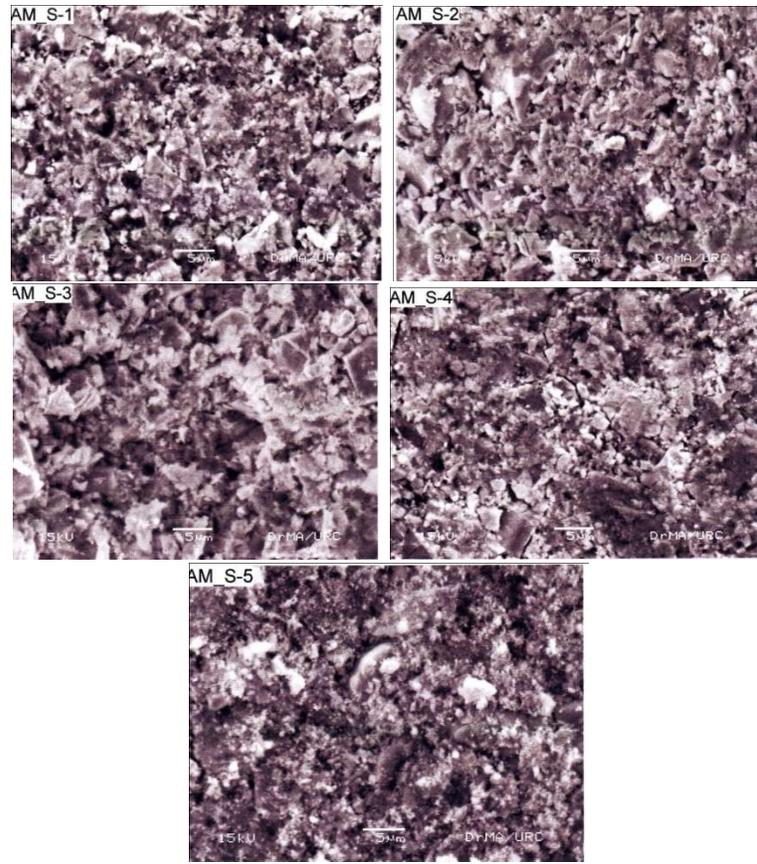


Figure 1.2: The SEM image of wolframite samples with different annealing temperatures.

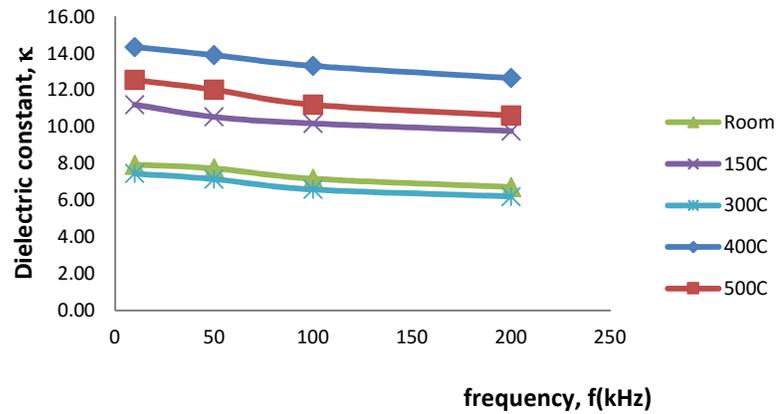
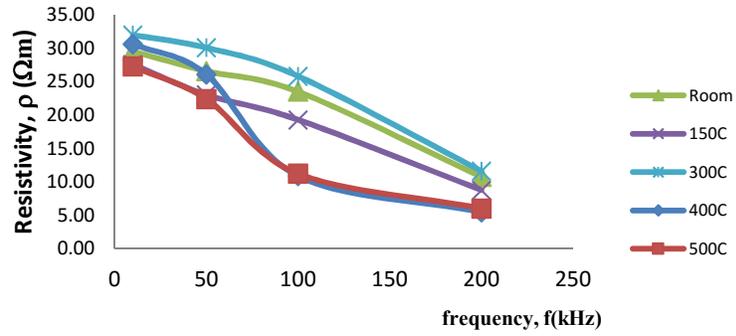
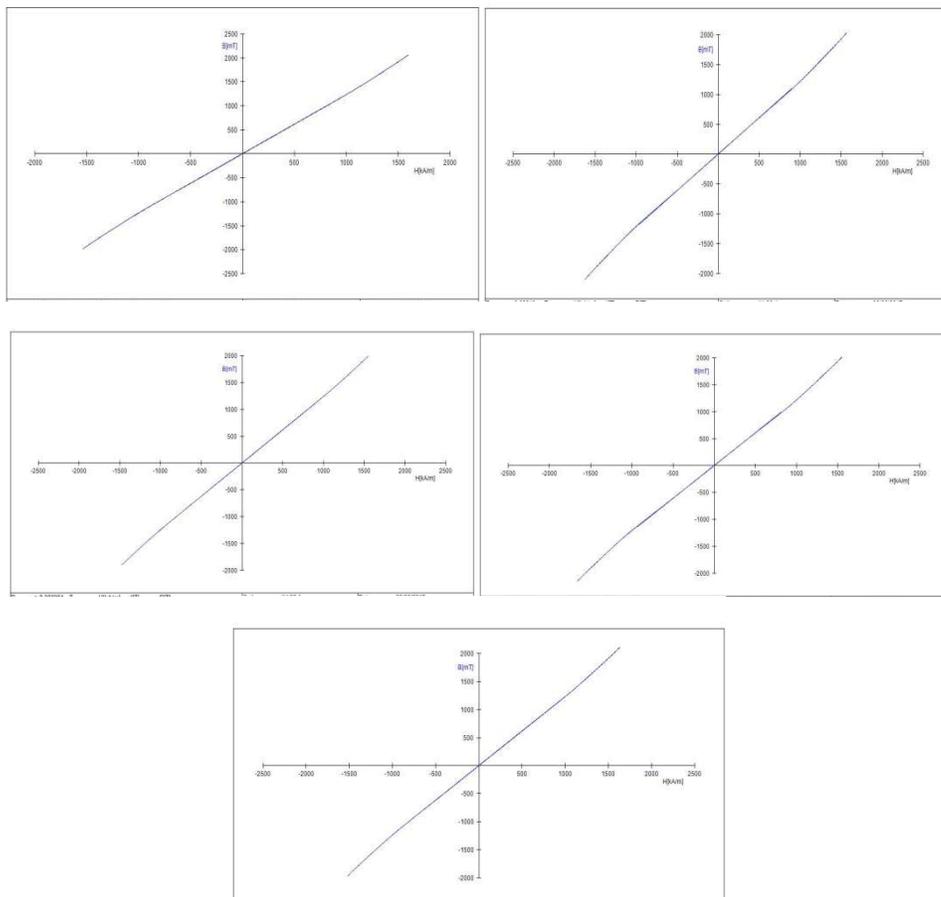


Figure 1.3: The capacitive effects of wolframite samples



**Figure 1.4:** The frequency dependence resistivity of wolframite samples.



**Figure 1.5:** The Magnetic properties of wolframite samples

## **Acknowledgements**

I would like to express appreciation to Pro-rector Dr Ba Han, Pro-rector Dr Khin May Aung, Dawei University and Pro-rector Dr Aung Myat Kyaw Sein, Mawlamyine University for their encouragement and kind permission to undertake the present research. I also would like to express my profound thanks to Professor Dr Than Tun, Head of Department of Physics, and Professor Dr Sanda Dwe, Department of Physics, Mawlamyine University, for their kind permission to use material science research lab apparatus in Mawlamyine University and Professor Dr Aung Myat Kyaw, Department of Physics, Dawei University for encouragement and help during this work.

## **References**

1. Ahmad S et al, (2004), *Ferroelectric Ceramics Processing Properties and Applications*, Department of Ceramic Science and Engineering, Rutgers University: USA  
Instruction Manual LCR meter GW 820, USA. JCM 6000 Plus, Scanning Electron Microscope.
2. Moulson A J & Herbert J M, (1997), *Electroceramic Materials Properties Applications*, New Delhi: Thomson Press
3. Rajagopal K. (2009) *Textbook of engineering physics part II*. New Delhi-110001.
4. Steingroever GmbH. Dr.,(2010), *Computer controlled permagraph® L*” Cologne-Germany
5. Suryanarayana C. and Norton M. G. (1998), *X-ray Diffraction A Practical Approach*, (New York: Plenum).
6. Tyagi M S, (1991), *Introduction to Semiconductor Materials and Devices*, New York: Wiley
7. Xu Y, (1991), *Ferroelectric Materials and their Applications*, New York: Elsevier Science Publishing Co. Inc.
8. Yadav M S, (2003), *A Textbook of Spectroscopy*, New Delhi: Anmol Publication  
<http://www.digibridge-ab.ca/ceramic material/86.html>  
<https://en.m.wikipedia.org/wiki/Hysteresis>